









CONTROLLING THE MORPHOLOGY, CHEMISTRY, AND PROPERTIES OF NANOCARBONS IN THE NANOPRODUCT LAB

Catalytic CVD of Carbon Nanotubes (CNTs)

Typically gas-phase decomposition of hydrocarbon precursor, catalyst nanoparticle formation by thin film dewetting, and catalytic surface reactions in CNT nucleation/growth are coupled.



Why decouple? 1. It's difficult to fully understand each process independently 2. It's more challenging to optimize growth in coupled recipes

Golnaz Tomaraei¹, Soumalya Ghosh¹, Mirza Sahaluddin¹, Moataz Abdulhafez¹, KiHo Nam¹, Jaegeun Lee¹, Mostafa Bedewy^{1,2,3}

¹NanoProduct Lab, Department of Industrial Engineering, Swanson School of Engineering, University of Pittsburgh, PA ²Department of Chemical and Petroleum Engineering, Swanson School of Engineering, University of Pittsburgh, PA ³Department of Mechanical Engineering and Materials Science, Swanson School of Engineering, University of Pittsburgh, PA mbedewy@pitt.edu

Laser Induced Nanocarbon (LINC)

Laser carbonization of polymers is an emerging technique that enables directly patterning conductive carbon electrodes for a plethora of flexible devices, including supercapacitors and sensors.

A continuous beam with power P is scanned across the polyimide film at a speed v with the sample surface at a distance z from the beam waist.

Photothermal interactions from the radiation absorption by the polyimide drive a rapid temperature increase that carbonizes polyimide locally and forms nanoscale sp^2 nanocarbon.





DSDA-OD



Fluence dependent heteroatom-doped LINC electrodes with comparable carbon

Custom-designed RTP-CVD reactor for decoupling

Custom-designed reactor: resistive preheater for decomposition of gas-phase hydrocrabon and IR furnace (heating rate > 200 °C/s) for catalyst treatment and CNT growth



Decoupling T_p, T_c, T_g in this RTP-CVD reactor

More than 12-fold increase in forest height by heating the gases as they go through the spiral gas injector above 800 °C



Decoupled recipe \rightarrow temp. is ramped in a separate catalyst formation step before the CNT growth step

Preheater --- H₂O bubbler 1000 C_2H_4 800 08 1200 Catalyst 1000 CNT growth 600 formation 800 400 600 80 100 120 40 60 Time [min]





Fluence gradients for control of LINC Morphology

Tilting the sample allows scanning the sample with different fluence values and hence investigate how fluence affects morphology of LINC. Discrete morphological transitions noted.





NSF award number 2028580 | NSF Award number 1825772 | Pitt Momentum Seed Grant

Density & Height Control of VACNT in this Reactor

I. CNT forest height is directly proportional to catalyst formation temp. (T_c) at the same growth temp. (T_g) .



II. CNT forest density is inversely proportional to growth temp. (T_q) , regardless of the catalyst formation temp. (T_c) .



Density is nearly independent on T_c , suggesting that catalytic activation density is mainly dependent on T_g

